

Investigation of SO₂ Adsorption of Graphene Oxide Fiber Bundle

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Abstract—This study focuses on the SO₂ adsorption properties of graphene oxide fiber bundle. Graphene oxide fiber bundle was produced by NaOH reduction and Hummer's method, and SO₂ adsorption tests were carried out. According to the SEM analysis and EDS Spectra, sulphur particles accumulated on the surface of the graphene oxide fibers were observed, in along with the result of the EDS spectra showing 6.7% sulphur content on the graphene oxide fibers. The absorption was also 102 mg of SO₂ per gram of sorbent (mg/gram) according to titration test.

Keywords—graphene oxide fiber bundle, SO₂ adsorption

I. Introduction

One of the most hazardous gases people need to cope with is SO₂, which is an industrial waste, that can cause acidic rains in along with greenhouse effect [1-3]. Gas sensors to detect SO₂ and pollution control is the current focus of many researches; however, have a limitation of high temperature requirement [1,2]. Graphene, due to its high surface area, is thought to increase the amount of gas absorbed, and might give rise to the low-weight room temperature sensor applications [1]. The possibility of graphene, detecting various gasses have been reported by several researchers [3-5].

This study focuses on the SO₂ adsorption and desorption properties of graphene oxide fiber bundle. The aim of the study is to investigate the SO₂ adsorption properties of graphene oxide (GO) fiber bundle produced as a nonwoven film structure for possible applications on gas detectors and filtration. In the current literature, air and water filtration properties of graphene sheets and graphene coated fibers are studied; however, there is no source that can show the possibility of graphene oxide fibers for air filtration and gas detectors. With this study, we aim to develop of graphene oxide fiber based systems with SO₂ adsorption properties for filtration and improve the gas filtration performance by the high surface area, and functional groups on the surface-

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II. Materials and Methods

A. Materials

Nonwoven films were manufactured from reduced graphene oxide (GO) fiber bundles (Natural Flake Graphite, with 300 μm fiber diameter and 20 μm thickness). GO was prepared according to Hummer's method. Nonwoven structures were obtained from fibers drawn from GO dispersion in coagulation baths of containing 5% NaOH by weight and water/ethanol solutions.

B. Methods

SO₂ adsorption and desorption tests were carried out at quartz reactors on at the lab scale testing mechanism. SEM images and EDS spectroscopy was performed under SEM microscopy after SO₂ adsorption.

SO₂ adsorption tests were carried out with the GO fibers in the nonwoven form. The nonwoven surfaces were placed into the tubular oven inside the quartz reactor. Adsorption process was conducted at room temperature for 2 hours by feeding 150 ml/min of N₂ and 1.4 ml/min SO₂ gasses to the system. GO fibers were subjected to desorption process with the same system used for adsorption tests. Desorption process was conducted by the stepwise increase of the temperature at an interval of 5 °C/min up to 360 °C for an hour. N₂ gas feeding to the system with a volume of 150 ml/min was continued in desorption process. During desorption, the amount of SO₂ adsorbed on GO nonwoven surface was determined by titration. The desorbed SO₂ gas was accumulated by a gas-washing bottle, which contained 100 ml of H₂O₂ (5% by wt). At the end of the process, the amount of H₂SO₄ on the bottle was determined by titration and adsorbed SO₂ amount was calculated from the H₂SO₄ amount.



Figure 1. SO₂ adsorption and desorption testing mechanism

III. Results and Discussion

Fibers were coagulated in the coagulation bath for 2 minutes and has crimped structure with shrinkage. As a result of NaOH usage, the fibers were partially reduced and obtained a darker color as can be seen in Figure 2.

It was observed that the fibers adsorbed 102 mg of SO₂ sorbent gram (mg/g). The GO fiber bundle before and after the adsorption process is shown in Figure 3. As can be seen from the Figure 3, fibers shrunk and became more brittle after the desorption process at 360 °C.

The pH level of the gas washing solution was measured as ph 1-2 after desorption. This might give rise to thought that absorbed SO₂ on the sample might have successfully removed from the surface and accumulated on the gas washing solution. Figure 4 shows the gas washing solution before and after titration process, and as seen, the color of the solution changed from yellow to blue that proves the property of adsorption of SO₂ gas was gained.



Figure 2. Change in fiber color after partial reduction



Figure 3. The a)pre-adsorption and b) after desorption images of nonwoven surfaces obtained from coagulation in pH 7 NaOH containing coagulation bath

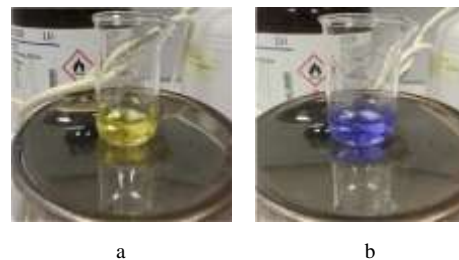


Figure 4. The gas washing solutions a) before titration b) after titration

After adsorption step, before the nonwoven surfaces proceed through desorption, SEM analysis in along with EDS spectra were carried out and the sulphur on the surface was proved. As can be seen from Figure 5, after SO₂ adsorption, sulphur particles were accumulated on the surface and the amount of sulphur was found as 6.7% by EDS spectra shown in Figure 6.

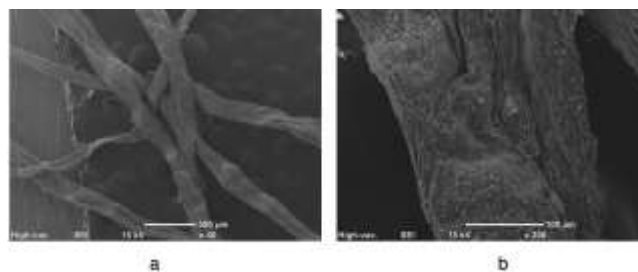


Figure 5. Fiber morphology after adsorption a-fiber structure, b- surface structure

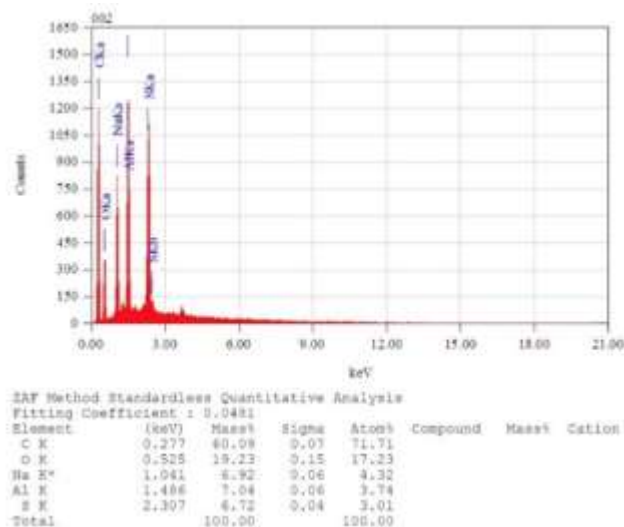


Figure 6. EDS spectra of nonwoven surfaces after adsorption

IV. Conclusions

To determine the possibility of use of graphene oxide fibers in gas sensors and detectors and filtration, the SO₂ adsorption of graphene oxide nonwoven surfaces was investigated. Graphene oxide nonwoven surfaces was prepared by using Hummer's method continued with a NaOH reduction system. Shrinkage and brittleness of the fibers were observed. The amount of SO₂ adsorbed on the fiber surface was found to be 6.7% according to EDS spectroscopy and sulphur particles on the surface was also observed in the SEM images. After the desorption process, the removal of the SO₂ from graphene oxide nonwoven surface to gas washing solution was determined 102 mg of SO₂ per gram of solvent (mg/gram) was found to be adsorbed by the graphene oxide nonwoven surface.

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