PHYTOCHEMICAL INVESTIGATION AND PHARMACOLOGICAL EVALUATION OF CURCUMIN BY SPEC-UV-VISIBLE SOFTWARE TECHNIQUE

[Charu, Rajan Chaudhary]

Abstract— Spectral analysis of the isomers of curcumin was studied using Spec-UV Software, for peak picking, graphics printout, Absorbance ratio calculation for selected wavelengths which included MPU Software Platform/Spec UV software workstation and spectral bandwidth of 2nm. Turmeric, a food additive, which has a main component curcumin, has shown surprisingly beneficial effects in experimental studies in the diseases characterized by an inflammatory effects. Several natural substances have greater antioxidant effects, including and curcumenoids. The extract of the curcumin was prepared and chemical tests were performed with structural analysis. The pharmacological properties as antioxidant effects were studied.

Keywords—Spec-UV software, curcumin, antioxidant

Introduction

Turmeric (Curcuma longa) is a member of the curcuma group, which is part of the ginger family of herbs. The root and rhizome (underground stem) of the turmeric plant is crushed into ground turmeric. Curcumin is medically promising because inflammation [1] and oxidative damage [2] are contributors to so many diseases as Alzeimer's [3], Parkinson [4], arthritis [5] and various cancer [6]. It is a potent antioxidant [7] with specific antiviral, anticancer and cholesterol lowering [8] effects. Curcumin has potent irreversible antiproliferative effect against a variety of cancer cell lies in vitro. It is a mixture of three related compounds extracted from the roots of curcma longa plants, i.e, curcumin, demethoxycurcumin and bisdemethoxycurcumin [9]. The active substance of turmeric is the polyphenol curcumin also known as natural yellow3 [10].

¹Charu

¹Dept. of Chemistry, Future Institute of Engineering and Technology, Bareilly India

²Rajan Chaudhary ²Dept. of Electronics and Communication, Future Institute of Engineering and Technology, Bareilly India, Systematic name of curcumin is 1,7-bis (4-hydroxy-3methoxyphenyl)-1,6-heptadiene-3,5-dione [11]. It exists in two tautomeric [12] forms keto in solid phase and enol in solution. The yellow pigmented fraction isolated from the rhizomes of curcuma longa contains curcuminoids present to the extene of 3-5% which is an important active ingredient responsible for the biological activity Spec-UV Software is used for structural analysis and patent designing of the isomers of curcumin whose special features are Log Record, Binary File Save, Multi-User Management, Quality Control, Software Conforming to GLP, Data Printout, Printout Records.

п. Experiment

Extraction of curcumin from rhizomes of turmeric was performed.100 gm of turmeric was crushed in a grinder and extracted in 95% alcohol in a soxhlet assembly until all the colouring matter was extracted [14]

The alcoholic extract was distilled off to a semisolid brown coloured mass about 4.5%. The crude extract was than dissolved in 100ml of benzene and extracted twice with equal volume of 0.1% NaOH. Alkaline extract are combined and acidified with dilute HCl, a yellow coloured compound was formed, which was concentrated on a water bath and at the same time the precipitate was dissolved in boiling water until a lumpy mass was formed. The solution was then filtered in hot condition and filtrate was concentrated to very small volume and finally cooled to get curcumin 1.5%.

III. Chemical Test for Curcumin

- 1. Curcumin when dissolved in concentrated sulphuric acid gives yellow –red colouration.
- 2. Curcumin when dissolved in 0.1 NaOH gives deep brown colour.

Isolation of the isomers: TLC method was used containing petroleum ether and methanol in the ratio of 1:3, left for 24 hrs. and PET was used to observe the spots. It was observed that Rf calues for isomers were 0.49, 0.45, 0.54, 0.55 tetra hydrocurcumin, demethoxycurcumin, tetra hydrodemethoxycurcumin, bisdemethoxycurcumin.

IV. Results

The authors observed that bisdemethoxycurcumin was less susceptible to degradation at pH 10.2 than

DIGITAL LIBRARY

demethoxycurcumin [15]. It recomended was that bisdemethoxycurcumin was used in alkaline composition improve stability. In serum free buffer solutions curcumin was observed to decompose in a pH dependent manner with faster reactions at neutral basic conditions. The authors observed intermolecular hydrogen bonding in both the ground and excited states. The curcuminoids possesing a phenolic group, which showed hydrogen bond acceptor properties, while those in bisdemethoxycurcumin acted as hydrogen bond donors, explaining the differential polarity of these curcumins when mixed with different alcohols. The comparative studies antioxidant activity of curcuminoids tetrahydrocurcumin in vitro using linoleic acid as a substrate in ethyl alcohol -water system as well as using rat liver showed that tetrahydrocurcumin had the strongest activity among all the curcuminoids in assay system. The authors concluded that these results suggest that tetrahydrocurcumin must play an important role in the antioxidant mechanism of curcumin in vivo.

The results have been illustrated in figures given below:

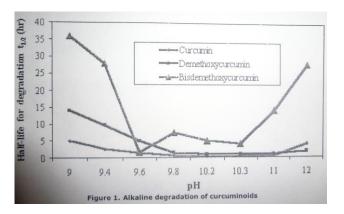


Figure 1. Alkaline degradation of curcuminoids.

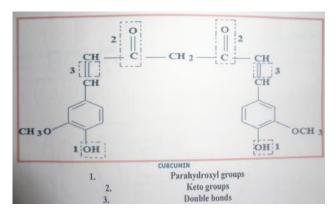


Figure 2. Structure of curcumin.

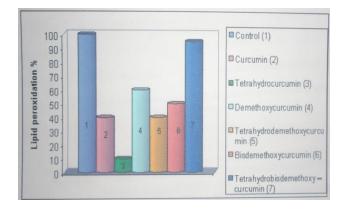


Figure 3. Comparative Antioxidant Activity of the Curcuminoids and Tetrahydrocurcuminoids in rat erythrocyte membrane system model (determined by TBA method)

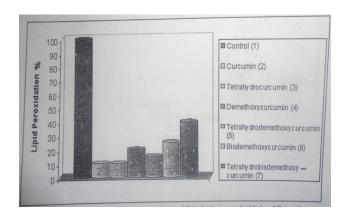


Figure 4. Comparative Antioxidant Activity of the Curcuminoids and Tetrahydrocurcuminoids in rat liver microsomes model (determined by TBA method)

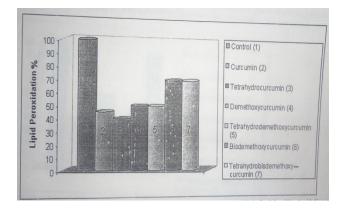


Figure 5. Comparative Antioxidant Activity of the Curcuminoids and Tetrahydrocurcuminoids in linoleic acid autoxidation model (determined by TBA method)



$$R_{3}$$
 R_{4}
 R_{5}
 R_{5}
 R_{6}
 $R_{1,7-bis(R-Phenyl)-1, 6-heptadiene-3, 5-dione (in enol form)$

Figure 6. General structure.

Spectroscopic data [16-20] of curcumin are shown in spectroscopic graphs using Spec -UV software for the analysis of the polyphenolcurcumin at different wavelenghths.

The T80 series of UV-Visible Spectrophotometers carry on various spectrum scans and quantitative determination. When interfaced to a PC, the software offers many more user-friendly applications as access to data base, three-dimensional spectrum analysis. Multi-user management, Log record, as well as quality control and printout record are the main features of the software.

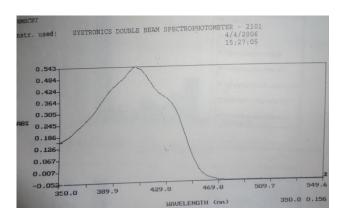


Figure 7. Absorbance of curcumin studied by UV-Visible spectrophotometer for peak 1.

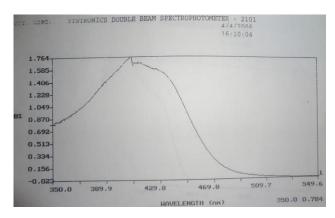


Figure 8. Absorbance of curcumin studied by UV-Visible spectrophotometer for peak 2.

References

- Shishodia S, Sethi G, Aggarwal BB. Getting back to roots. Ann NY Acad Sci. 2005;1056:206–17.
- [2] Frautschy SA, Hu W. Phenolic anti-inflammatory antioxidant reversal of b induced cognitive deficits and neuropathology. Neurobiol Aging. 2001;22:993–1005.
- [3] Ammon HP, Wahl MA. Pharmacology of curcuma longa. Planta Med. 1991;57:1–7.
- [4] Di Patre PL, Read SL, Cummings JL, Tomiyasu U, Vartavarian LM, Secor DL, et al. Progression of clinical deterioration and pathological changes in patients with Alzheimers Disease evaluated at biopsy and autopsy. Arch Neurol. 1999;56:1254–61.
- [5] Fiala M, Liu PT, Espinosa-Jeffrey A, Rosenthal MJ, Bernard G, Ringman JM, et al. Innate immunity and transcription of MGAT-III and Toll-like receptors in Alzheimers disease patients are improved by bisdemethoxycurcumin. Proc Natl Acad Sci USA. 2007;104:12849–54.
- [6] Masuda, T., Maekawa, T., Hidaka, K., Bando, H., Takeda Y. and Yamaguchi, H., Chemical studies on antioxidant mechanisms of curcumin: analysis of oxidative coupling products from curcumin and linoleate. Journal of Agriculture and Food Chemistry 49, 2539–2547, 2001.
- [7] Youssef KM, El-Sherbeny MA. Synthesis and antitumor activity of some curcumin analogs. Arch Pharm (Weinheim) 2005;338:181–9.
- [8] Ireson CR, Jones DJ, Orr Sl. Metabolism of the cancer chemopreventive agent curcumin in human and rat intestine. Cancer Epidemiol Biomarkers Prev. 2002;11:105–11.
- [9] Puglielli L, Tanzi RE, Kovacs DM. Alzheimer's disease: The cholesterol connection. Nat Neurosci. 2003;6:345–51. [PubMed]
- [10] Soni KB, Kuttan R. Effect of oral curcumin administration on serum peroxides and cholesterol in human volunteers. Indian J Physiol Pharmacol. 1992;36:273–5.
- [11] S. Li, W. Zhu, Z. Xu, J. Pan, H. Tian, Tetrahedron 2006, 62, 5035
- [12] T. Kojima, S. Miyazaki, K. Hayashi, Y. Simazaki, F. Tani, Y. Raruta, Y. Manda, Chem. Eur. J., 2004, 10, 6402
- [13] Chainani N. Safety and anti-inflammatory activity of Curcumin component of turmeric (curcuma longa) J Alter Compl Med. 2003:9:161–8.
- [14] On the antioxidant mechanism of curcumin: classical methods are needed to determine antioxidant mechanism and activity. Org. Lett., 2(18): 2841-3.
- [15] Stability of curcumin in buffer solutions and characterization of its degradation products. J. Pharm. Biomed. Anal., 15(12): 1867-1876.
- [16] Studies of curcumin and curcuminoids: Alkaline degradation of curcumin. Z. Lebensm. Unters. Forsch. 180: 132-134 Wan, Y.J., Pan, M.H., Cheng, A.L., Lin, L.I., Ho, Y.S., Hsieh, C.Y., & Lin, J.K. 1997
- [17] Kolev TM, Velcheva EA, Stamboliyska BA, Spiteller M. DFT and experimental studies of the structure and vibrational spectra of curcumin. Int J Quantum Chem. 2005;102:1069–79.
- [18] Proliferation and growth. Neuro Endocrinol Lett. 2003;24:469–21.Srinivasan, K.R. 1952. The colouring matter in turmeric. Current Science, 311-313.
- [19] Sugiyama, Y., Kawakishi, S. & Osawa, T.. 1996. Involvement of the beta-diketone moiety in the antioxidative mechanisms of tetrahydrocurcumin.
- [20] Biochem. Pharmacol. 52(4): 519-525 Pabon, H. J. J., 1964. A synthesis of curcumn and related compounds. Rec. Trav. Chim. Pays-Bas. 83:379-386 Tonnesen, H.H & Karlsen, J., 1985.



Proc. of the Intl. Conf. on Advances In Engineering And Technology - ICAET-2014 Copyright © Institute of Research Engineers and Doctors. All rights reserved. ISBN: 978-1-63248-028-6 doi: 10.15224/978-1-63248-028-6-02-39

About Authors: Dr.Charu, Associate Professor PHD. 14 years experience in renouned institutes.



Er. Rajan Chaudhary Assistant Professor, M.Tech (ECE) Author of 21 international research papers till now.



Spectral analysis of the isomers of curcumin was studied using Spec-UV Software, for peak picking, graphics Absorbanceprintout, ratio calculation for wavelengths selected which included MPU Software Platform/Spec UV software workstation and spectral bandwidth of 2nm.

